(54) PRODUCTION OF DICYANAMIDE METAL SALT

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ORPOSE: To produce a dicyanamide metal salt having high purity in high yield, by reacting ammonia with specific amounts of a cyanogen halide and a metal 2 7 0 0 0

CONSTITUTION: Ammonia is added to an aqueous solvent to a concentration of about $5\sim20\mathrm{wt}\%$ and 1 equivalent of the ammonia is made to react with 1.8 - 2.2 equivalent of a cyanogen halide and 2.8 - 3.3 equivalent of a metal hydroxide. The cyanogen halide is e.g. cyanogen chloride and the metal hydrox. ide is e.g. sodium hydroxide. The reaction is carried out at about 0~60°C for about 0.5 \sim 6hr. The produced dicyanamide metal salt is separated by filtration of the reaction mixture optionally after concentration. If necessary, the separated salt is purified by washing with cold water, etc. A high-purity dicyanamide hydroxide based on said ammonia in an aqueous solvent. metal salt can be produced at a yield of about ≥90%.

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